

Practical: Exercise 3. Hydrogeochemistry of Faryab

The four analyses below are from real groundwaters in Pas Sang (Kohistan), Qor Qol (Qaysar), Nawabad (Bilchiragh) and Tawachi (Andkhoi)

		Pas Sang	mmol/l	meq/l	Qor Qol	mmol/l	meq/l	Nawabad	mmol/l	meq/l	Tawachi	mmol/l	meq/l
pH		8.0			8.6			7.3			7.8		
Electrical Conductivity	$\mu\text{S/cm}$	356			940			2450			10070		
Na^+	mg/l	7.8			59.1			172			1469		
K^+	mg/l	0.92			6.08			5.88			26.0		
Ca^{++}	mg/l	55.3			67.5			88.1			218		
Mg^{++}	mg/l	7.06			60.5			189			580		
SO_4^-	mg/l	17.3			189			394			2486		
NO_3^-	mg/l	6.12			24.6			43.3			88.0		
Cl^-	mg/l	2.15			34.4			179			1919		
F^-	mg/l	0.13			0.20			0.87			1.81		
HCO_3^-	mg/l	198			331			781			846		

Convert the analyses first to mmol/l and then to meq/l (use the atomic masses provided).

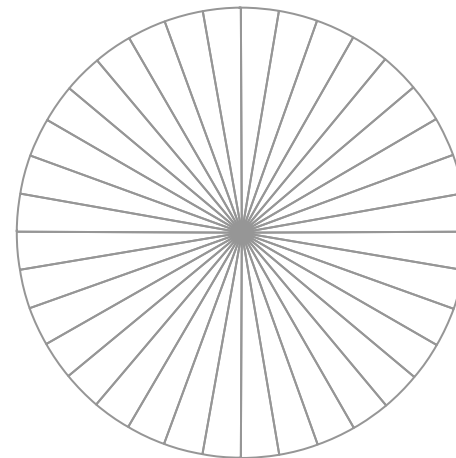
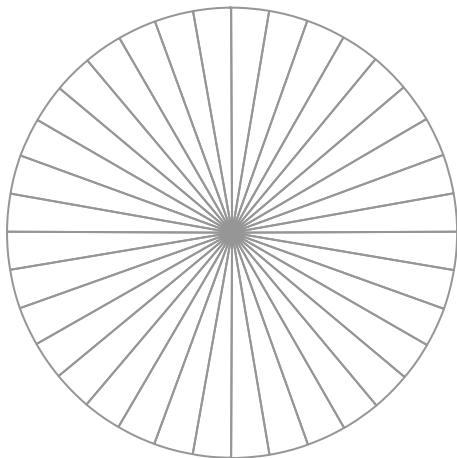
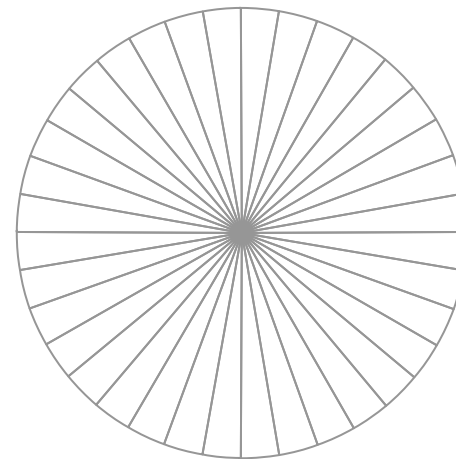
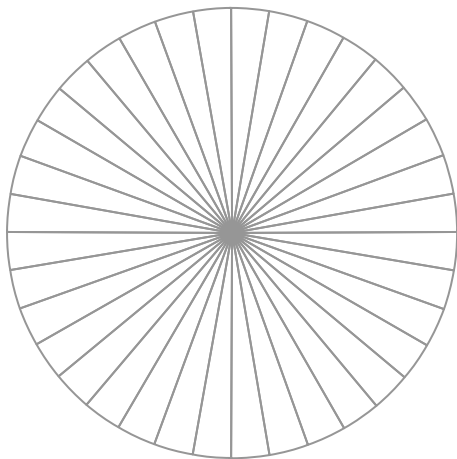
For each water, state the dominant cation and anion in terms of milliequivalents per litre. Calculate the ion balance.

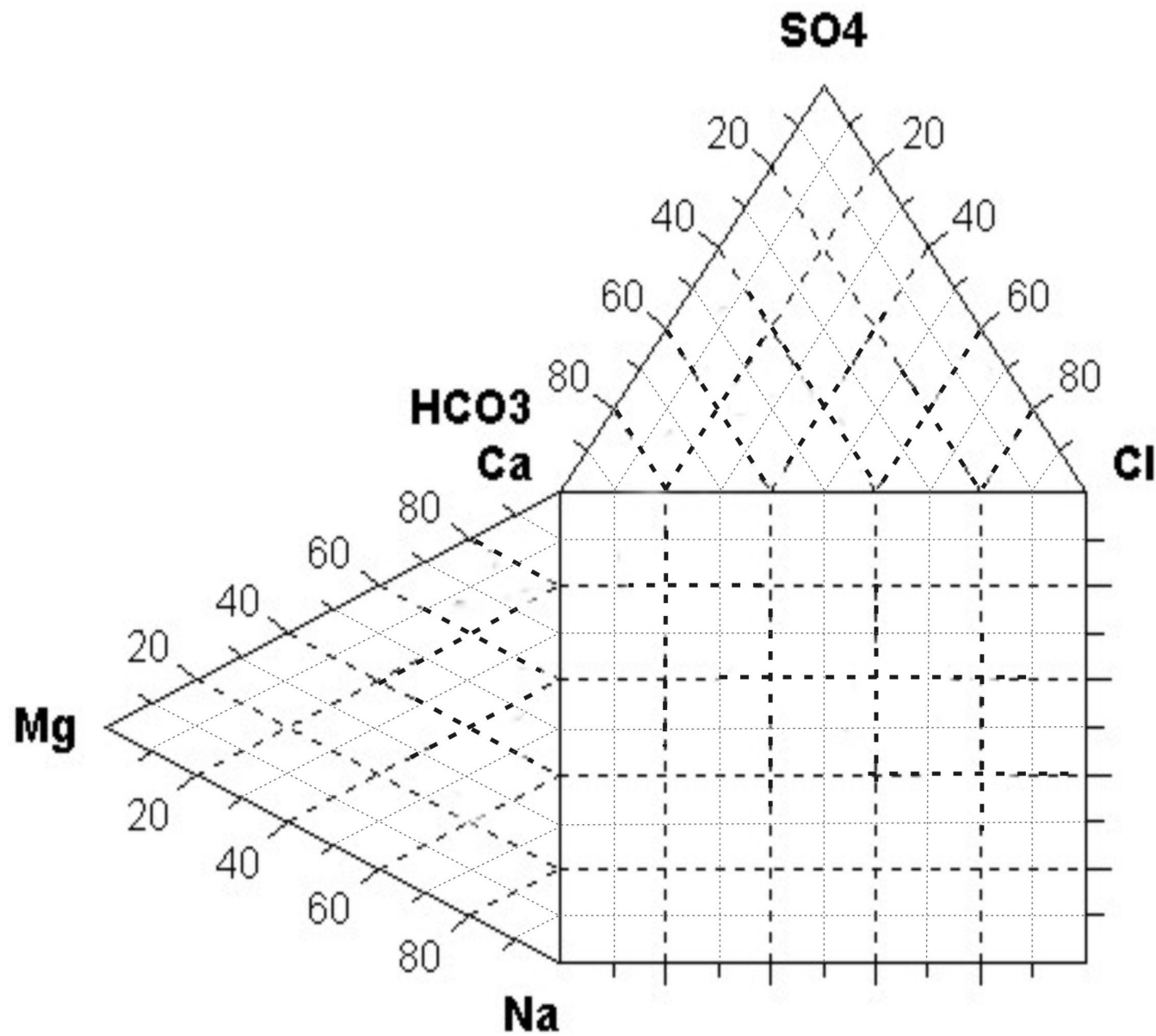
What are the main differences that you notice between the various samples?

Plot the analyses as pie diagrams, and on the Durov diagram

Symbol	Name	Atomic mass (g/mol)
N	Nitrogen	14.007
O	Oxygen	15.999
F	Fluorine	18.998
Na	Sodium	22.99
Mg	Magnesium	24.305
S	Sulphur	32.066
Cl	Chlorine	35.453
K	Potassium	39.098
Ca	Calcium	40.078

Note: $\text{pH} = -\log_{10} (\text{H}^+)$, where (H^+) is the activity of hydrogen ions in moles per litre.
 To convert mg/l to mmol/l: $\text{Conc (mmol/l)} = \text{Conc (mg/l)} / \text{atomic mass}$
 To convert mmol/l to meq/l: $\text{Conc (meq/l)} = \text{Conc (mmol/l)} \times \text{Ionic Charge}$





Durov diagram